

## ABSTRACT:

### H<sub>2</sub> Chemical Storage in E-Fuels: Novel Materials for Power-to-X

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The urgent need to mitigate anthropogenic CO<sub>2</sub> emissions while ensuring energy security has positioned Carbon Capture Utilisation and Storage (CCUS) technologies as a benchmark for the global energy transition [1]. In the CCU framework, this work reports recent results on CO<sub>2</sub> catalytic hydrogenation into high-value energy carriers via Power-to-Methane (PtM) and Power-to-Liquid (PtL) pathways. Integrating advanced catalyst synthesis, multi-scale characterisation, and process simulation, it addresses coupling catalyst architecture and process design to overcome kinetic/thermodynamic barriers in circular carbon systems. For PtM, CO<sub>2</sub> methanation under steady-state and ICCU strategies is reported. Ni on USY zeolites showed interesting performance with coke resistance; hierarchical porosity and particle size/dispersion drove the catalytic behavior [2]. Rhenium dual-functional materials decoupled capture/conversion, achieving 100% methane selectivity via site-adsorbent synergy [3]. Conversely, PtL (CO<sub>2</sub> to methanol) faced distinct constraints: Cu catalysts are selective only at 150-220°C, beyond which RWGS/methanation dominate [4]. Metal-support interfaces can be tuned via proper selection of precursors. Cu-Fe-Ce/Cu-Fe screening at high pressure favored CO/CH<sub>4</sub> over methanol, warranting further study. Selectivity depends on metal composition, redox, and conditions. Overall, effective CO<sub>2</sub> utilisation needs integrated catalyst-process design: PtM flexibility suits dynamic regimes; PtL demands precise catalysts + advanced reactors.

[1] International Energy Agency, World Energy Outlook 2025, 2025. [www.iea.org/terms](http://www.iea.org/terms).

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